

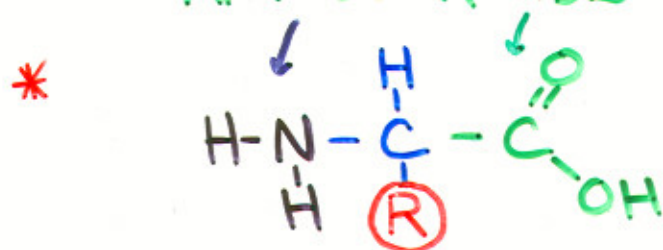
# C. PROTEINS

## 1. FUNCTION:

- **STRUCTURAL** eg. KERATIN - HAIR + NAILS, COLLAGEN - CONNECTIVE TISSUE
- **ENZYMES** - SPEED UP CHEMICAL RXNS.  
SPECIFIC SHAPES REQUIRED
- **HORMONES** - CHEM. MESSENGERS (SOME)

## 2. STRUCTURE:

AMINO ACIDS = MONOMERS (a.a.)



AMINE

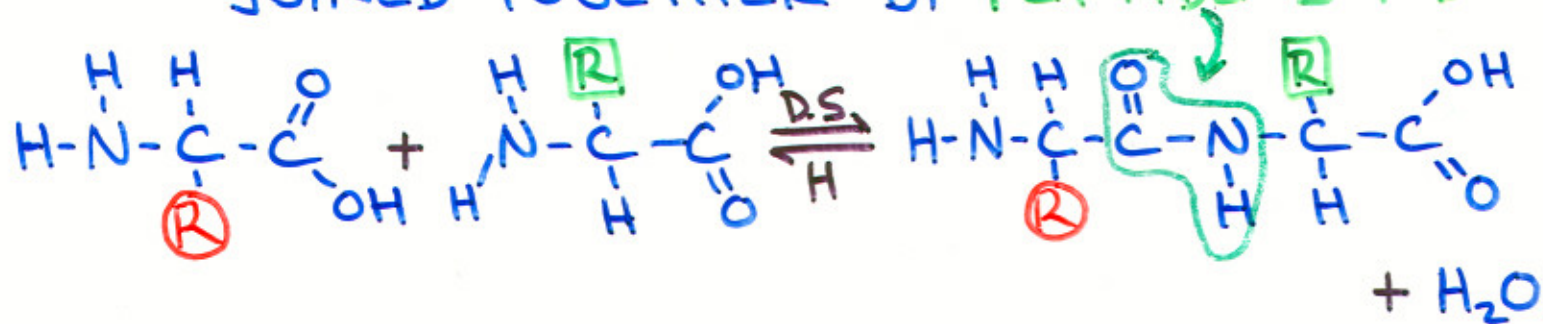
CARBOXYL

VARIES, DEPENDING ON AMINO ACID

20 DIFFERENT  $\textcircled{R}$  = 20 DIFF. a.a.

SEE Pg. 29/37

JOINED TOGETHER BY PEPTIDE BOND

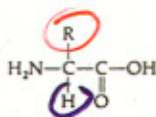


a.a. + a.a.  $\rightleftharpoons$  DIPEPTIDE +  $\text{H}_2\text{O}$

...  $\rightleftharpoons$  POLYPEPTIDE (100's  $\rightarrow$  1000's of aa)  
 $\hookrightarrow$  MACROMOLECULE

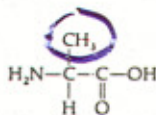
(a) Every amino acid contains an amino group ( $\text{NH}_2$ ) and a carboxyl group ( $\text{COOH}$ ) bonded to a central carbon atom. A hydrogen atom and a side group are also bonded to the same carbon atom. This basic structure is the same in all amino acids. The "R" stands for the side group, which is different in each kind of amino acid. (b) The 20 kinds of amino acids used in making proteins. As you can see, their basic structures are the same, but they differ in their side groups. The side groups may be nonpolar (with no difference in charge between one zone and another), polar but with the two charges balancing one another out so that the side group as a whole is uncharged, negatively charged, or positively charged. The nonpolar side groups are not soluble in water, whereas the charged and polar side groups are water-soluble.

(a)

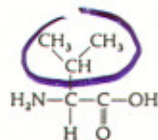


(b)

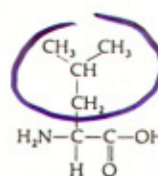
NONPOLAR



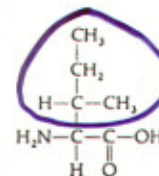
ALANINE (ala)



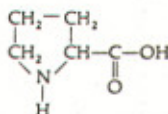
VALINE (val)



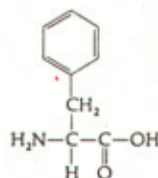
LEUCINE (leu)



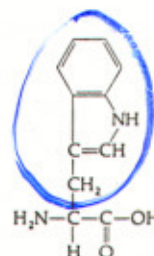
ISOLEUCINE (ile)



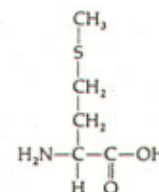
PROLINE (pro)



PHENYLALANINE (phe)

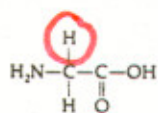


TRYPTOPHAN (trp)

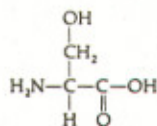


METHIONINE (met)

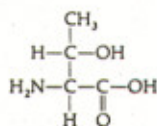
## POLAR BUT UNCHARGED



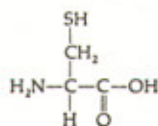
GLYCINE (gly)



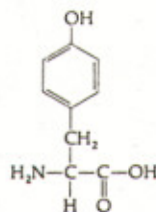
SERINE (ser)



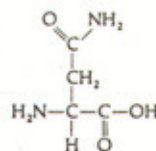
THREONINE (thr)



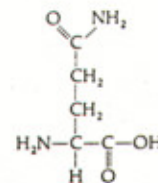
CYSTEINE (cys)



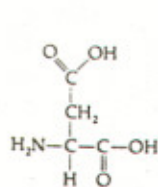
TYROSINE (tyr)



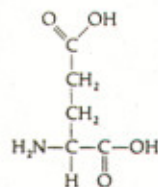
ASPARAGINE (asn)



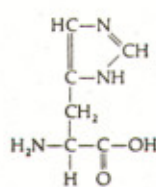
GLUTAMINE (gln)

ACIDIC  
(NEGATIVELY CHARGED)

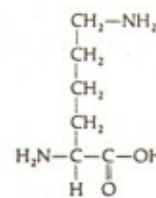
ASPARTIC ACID (asp)



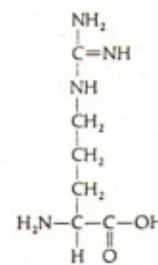
GLUTAMIC ACID (glu)

BASIC  
(POSITIVELY CHARGED)

HISTIDINE (his)



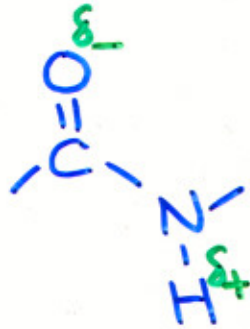
LYSINE (lys)



ARGININE (arg)



THE PEPTIDE BOND, LIKE  $H_2O$ , CARRIES PARTIAL CHARGES, ie. IS POLAR



HELPS GIVE PROTEINS THEIR SHAPE ...

4 DIFFERENT LEVELS OF ORGANIZATION  
SEE H.O.

DENATURATION - IRREVERSIBLE CHANGES  
IN SHAPE DUE TO EXTREMES IN **HEAT**  
AND **pH**

LOSS OF SHAPE = LOSS OF FUNCTION

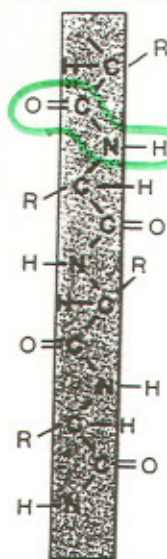
### 3. TYPES

- i. COMPLETE PROTEINS - ANIMAL PROTEINS  
CONTAIN ALL ~~8~~9 ESSENTIAL a.a.
- ii. INCOMPLETE PROTEINS - PLANT PROTEINS.  
LACKING AT LEAST 1 ESSENTIAL a.a.



# PROTEIN LEVELS OF ORGANIZATION P.39

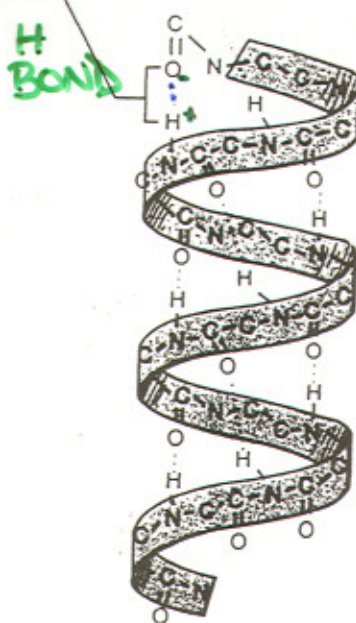
## 1. PRIMARY STRUCTURE



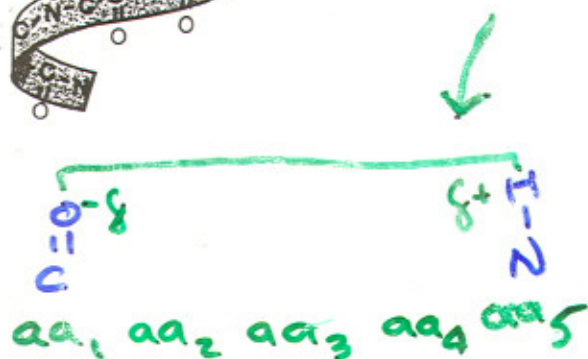
- LINEAR SEQUENCE OF A.A., JOINED BY PEPTIDE BONDS.

- DETERMINED BY DNA

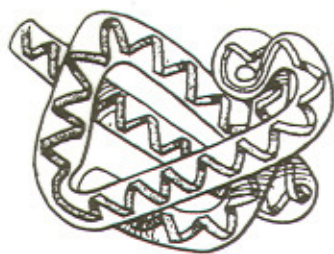
## 2. SECONDARY STRUCTURE



- $\alpha$  HELIX
- RIGHT HANDED
- 3.6 aa/TURN
- \* STABILIZED BY H BONDS



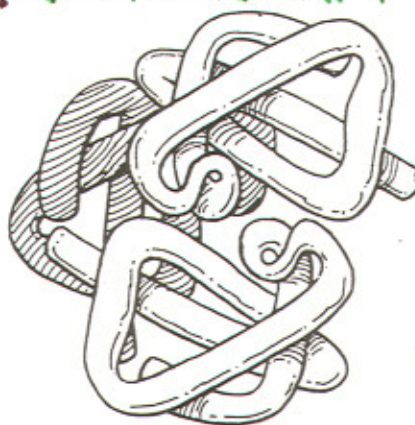
## 3. TERTIARY STRUCTURE



- FINAL 3D SHAPE
- GLOBULAR PROTEINS

- MAINTAINED BY IONIC, COVALENT + H BONDING BETWEEN  $R$  GROUPS

## 4. QUATERNARY STRUCTURE



- 2 OR MORE POLYPEPTIDES LINKED TOGETHER
- eg. HEMOGLOBIN